

*Vegetable oil as fat replacer inhibits formation of heterocyclic amines and 2 polycyclic aromatic hydrocarbons in reduced fat pork patties*

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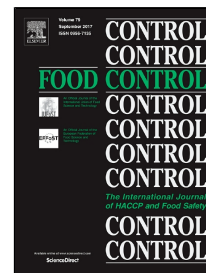
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# **Vegetable oil as fat replacer inhibits formation of heterocyclic amines and polycyclic aromatic hydrocarbons in reduced fat pork patties**

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## **Abstract**

Formation of heterocyclic amines (HCAs) and polycyclic aromatic hydrocarbons (PAHs) was examined to evaluate the impact of using vegetable oil as fat replacement on carcinogen formation in meat product. Pork patties were formulated with 40% fat replacement by olive oil, sunflower oil or grape seed oil, respectively and cooked at 180°C or 220°C. Control patties contained the highest amount of HCAs compared with all other patties at both temperatures. Olive oil and sunflower oil replacement completely inhibited formation of MeIQ (2-amino-3, 4-methylimidazo[4,5-f]quinoline), while grape seed oil completely inhibited MeIQx (2-amino-3,8-dimethylimidazo[4,5-f]quinoxaline), 4,8-DiMeIQx (2-amino-3,4,8-trimethylimidazo[4,5-f]quinoxaline) and PhIP (2-amino-1-methyl-6-phenylimidazo[4,5-b]pyridine) in patties. Grape seed oil achieved the highest inhibition capacity compared with sunflower oil and olive oil. HCAs increased significantly with cooking temperature ( $p < 0.05$ ), but no difference was observed in total PAHs for patties cooked at different temperature ( $p > 0.05$ ). In conclusion, fat replacement with sunflower oil, olive oil or grape seed oil in pork patties could reduce the formation of HCAs without compromising eating quality.

**Key words:** Antioxidants; Fat modification; Maillard reactions; Oxidation.

**Chemical compounds studied in this article:**

2-amino-3-methylimidazo[4,5-f]quinoline (PubChem CID: 53462); 2-Amino-3,4-dimethylimidazo[4,5-f]quinoline (PubChem CID: 62274); 2-Amino-3,8-dimethylimidazo[4,5-f]quinoxaline (PubChem CID: 62275); 2-amino-3,4,8-trimethylimidazo[4,5-f]quinoxaline (PubChem CID: 104739); 2-Amino-1-methyl-6-phenylimidazo[4,5-b]pyridine (PubChem CID: 1530); Benzo[a]pyrene (PubChem CID: 2336); Benz[a]anthracene (PubChem CID: 5954)

**1. Introduction**

Fat plays an important role in the human diet. It not only creates a unique sensation of food, but also helps maintain health. The consumption of pork in the world has dramatically increased from 18 to 110 million tons per year (1950-2010) (Brown, 2013). Research found that increased saturated fatty acid intake could elevate the risk of cardiovascular disease, but monounsaturated fatty acids (MUFA) and polyunsaturated fatty acids (PUFA) could reduce the risk and maintain cardiovascular health (McAfee et al., 2010; Sadler, 2014). Therefore, changing fatty acids profile of meat products by replacing saturated fatty acids with unsaturated fatty acids has attracted lots of attention in both academic research and meat processors. Adding olive oil could dramatically increase the percentage of MUFA in final products, whereas sunflower oil and grape seed oil could greatly raise the level of PUFA in fat replaced meat products (Gunstone, 2002; Matthäus, 2008). Rodríguez-Carpena et al. (2012) successfully replaced 50% fat with avocado, sunflower and olive oil in cooked pork patties and reported that avocado and olive oil could even offer better aroma to the final products than control ones. Vural and

Javidipour (2002) successfully substituted beef fat in Frankfurters with the mixture of interesterified palm, cottonseed, and olive oil without changing physical parameters and total sensory scores. Choi et al. (2010) used pre-emulsified grape seed oil and 2% rice bran fibre to develop pork batters with 50% fat replacement and reported that the fat-reduced pork batters could achieve the comparable eating quality with control samples. Domínguez, Agregán, Gonçalves, and Lorenzo (2016) replaced 100% pork back fat with olive oil in pork pâté, which significantly increase the content of tocopherol and MUFA in cooked products without altering physio-chemical properties. Domínguez, Pateiro, Agregán, and Lorenzo (2017) and Lorenzo, Munekeata, Pateiro, Campagnol, and Domínguez (2016) replaced 25%-75% backfat with olive oil, microencapsulated fish oil and the mixture of fish oil and olive oil, which significantly increased the percentage of PUFA in frankfurter type sausage and Spanish *salchichón*. These results indicate that vegetable oil could be used successfully to replace fat partially or completely to offer products comparable eating quality with healthier fatty acids profile, i.e. high level of MUFA and PUFA.

However, unsaturated fatty acids in vegetable oils may pose risk in domestic cooking due to their oxidation and decomposition at high temperature. For example, linoleic acid was found associated with the formation of potentially toxic compounds, such as free radicals, aldehydes and ketones (Guillén & Uriarte, 2012a; Katragadda et al., 2010). These reactive oxygen species (ROS) initiated by unsaturated fatty acids peroxidation could induce the decomposition of Amadori compounds and generate 1- and 3- deoxysone that are intermediates for Strecker aldehydes, pyrazines and pyridines in Maillard reaction. Consequently, it might promote the formation of heterocyclic amines (HCAs) (Morello, Shahidi & Ho, 2002; Turesky, 2010; Zamora & Hidalgo, 2007). Effect of fatty acids/oils on the formation of HCAs has been

documented in previous research. Johansson et al. (1995) reported that the higher level of MeIQx and DiMeIQx were found in burgers fried in rapeseed oil containing high level of oleic acid with high peroxides values, compared with butter, margarine and sunflower oil. Zamora et al. (2012) stated that both primary and secondary lipid oxidation products, hydroperoxides, such as methyl 13-hydroperoxyoctadeca-9,11-dienoate and alkenals could accelerate the formation of PhIP in chemical model system. Some hydroperoxides generated from the decomposition of the unsaturated hydrocarbons during heating, such as linolenate acid can also undertake aromatization and de-hydrocyclization, further cleave into benzaldehydes and other benzene ring-containing compounds, which are precursors of polycyclic aromatic hydrocarbons (PAHs) (Chen & Chen, 2001; Lorenzo et al., 2011; Lorenzo, Purriños, Fontán, & Franco, 2010; Singh, Varshney & Agarwal, 2016).

HCAs, PAHs and N-nitrous compounds are well-known carcinogens which were detected in processed meat products (Hasnol, Jinap & Sanny, 2014; Jinap et al., 2013; Liao et al., 2010; Oz & Kaya, 2010; Salmon, Knize & Felton, 1997). HCAs are mainly formed with the presence of free amino acids, carbohydrates and creatine under high cooking temperature (Rahman et al., 2014). IARC (1993) classified the following 5 aminoimidazoarenes (AIAs) compounds as human carcinogens, including 2-amino-3-methylimidazo[4,5-f]quinoline (IQ), 2-amino-3,4-methylimidazo[4,5-f]quinoline (MeIQ), 2-amino-3,8-dimethylimidazo[4,5-f]quinoxaline (MeIQx), 2-amino-3,4,8-trimethylimidazo[4,5-f]quinoxaline (4, 8-DiMeIQx) and 2-amino-1-methyl-6-phenylimidazo[4,5-b]pyridine (PhIP). PAHs are hydrocarbons that contain two or more benzene rings, such as pyrene, anthracene and naphthalene. They can be formed through incomplete combustion or pyrolysis of organic components, including fat, protein and carbohydrates at the temperature over 200 °C. Grilling, roasting and

smoking meat products likely contain high level of PAHs (Alomirah et al., 2011). Benz[a]anthracene (BaA) and benzo[a]pyrene (BaP) are the most potent carcinogenic PAHs in processed meat products (PHE, 2008). The metabolite of BaP, BaP-7,8-diol-9,10-epoxide, has been reported with the highest tumour-inducing activity due to causing DNA adducts (Purcaro, Moret & Conte, 2013).

Vegetable oils contain various antioxidants such as vitamin E,  $\beta$ -carotenes and phenolic compounds (Ramírez-Anaya et al., 2015). These antioxidants have been characterized as free radical scavengers during cooking, which might inhibit the formation of carcinogens (Janoszka, 2011; Wong, Cheng & Wang, 2012). Cheng, Chen and Wang (2007) reported that marinating beef patties with phenolic compounds such as epicatechin gallate, rosmarinic acids and carnosic acid could significantly reduce HCAs by 24%-70% in final cooked products. Balogh et al. (2000) found that HCAs (IQ, MeIQ, MeIQx, DiMeIQx and PhIP) were inhibited by 45%-75% when sprayed 1% vitamin E (w/w) on the surface of beef patties before frying. Therefore, in the concern of the carcinogen level in processed meat products, replacing saturated fat with vegetable oils rich in unsaturated fatty acids needs to be justified. Thus, the objectives of this study were to (1) explore the effect of partially replacing pork back fat with sunflower oil, olive oil and grape seed oil on the formation of HCAs and PAHs; (2) examine the effect of different cooking temperatures on the formation of carcinogens in fat reduced pork patties.

## **2. Material and methods**

### **2.1 Materials**

Three batches of lean pork leg and pork back fat with 40.3% SFA, 43.4% MUFA and 10.0% PUFA (McCance & Widdowson, 2002) were purchased from Jennings Caversham (Reading, UK) at different time point to consider the batch effect.



Excess visible fat on pork legs was trimmed, then minced by a Kenwood Food processor (Chef Titanium KM010, 4.6, Kenwood Limited) and vacuum packed separately. Raw materials were stored at -18 °C and defrosted 24h at 4 °C before use. Commercial grape seed oil (Waitrose®, produced in Italy) with 12.4% SFA, 20.2% MUFA, 68.2% PUFA, 10-15mg tocopherols and 5.9-11.5mg/100g polyphenols (Bail, Stuebiger, Krist, Unterweger, & Buchbauer, 2008), sunflower oil (Morrisons®, produced in UK) with 14.3% SFA, 20.5% MUFA, 63.3% PUFA and 50mg tocopherols (McCance & Widdowson, 2002) and refined olive oil (Filippo®, phenols were removed by industrial process, produced in Italy) with 14.3% SFA, 73.0% MUFA, 8.2% PUFA and 100-300mg tocopherols (McCance & Widdowson, 2002) were purchased from local supermarket (Reading, UK). Oils were kept in refrigerator (4°C) before making patties and further analysis.

The standards IQ (2-amino-3-methyl-imidazo [4,5-f] quinoline), MeIQ (2-amino-3,4-dimethyl-imidazo [4,5-f] quinoline), MeIQx (2-amino-3,8-dimthylimidazo [4,5-f]quinoxaline), 4,8-DiMeIQx (2-amino-3,4,8-trimethyl-imidazo [4,5-f] quinoxaline), PhIP (2-amino-1-methyl-6-phenylimidazo [4,5-b] pyridine, BaA (Benz[a]anthracene) and BaP (benzo[a]pyrene) were purchased from *Toronto Research Chemicals* (Toronto, Canada). Ammonium acetate, triethylamine, acetonitrile (HPLC grade), bovine serum albumin (BSA), dinitrophenylhydrazine (DNPH), ethyl acetate 99.5%, 0.9000g/ml, 6M guanidine HCl (pH 6.5), hydrochloric acid solution 0.1M, methanol (HPLC grade), HPLC grade water, sodium hydroxide 1M, perchloric acid (99.8%), sodium phosphate buffer (pH 6.5), thiobarbituric acid (TBA), and trichloroacetic acid (TCA) were purchased from *Fisher Scientific* (Loughborough, UK). 2,2-Azobis(2-methylpropionamidine) dihydrochloride granular 97% (ABAP), 2,2-Azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) diammonium salt (ABTS), (±)-6-Hydroxy-

2,5,7,8-tetramethylchromane-2-carboxylic acid (*Trolox*), phosphate buffer solution 0.1 M and phosphoric acid were purchased from *Sigma-Aldrich* (Gillingham, UK). The solid-phase extraction *Extrelut NT 20* columns and diatomaceous earth refill material were purchased from *Merck* (Darmstadt, Germany). Bond Elut propyl-sulfonic acid (PRS) cartridges (100 mg, 10 ml), C-18 cartridges (7 ml) were purchased from *VWR Inc* (Lutterworth, UK).

## 2.2 Procedures for preparing and cooking pork patties

The formulation of control patties (as shown in Table 2) included 700g lean pork mince, 180g distill water, 100g pork back fat and 20g salt per kilogram. For the fat partially replaced patties, 40% of pork back fat was replaced with sunflower oil, olive oil or grape seed oil respectively. Overall, there were 4 types of pork patties prepared in this study, i.e. control (100% back fat, C patties), sunflower oil patties (S patties), olive oil patties (O patties) and grape seed oil patties (G patties). All ingredients were homogenized at 5000rpm for 5min in the Kenwood Food processor to reach a uniform emulsion. Each patty was weighed 100g, shaped in a foil cup (9.0cm diameter \* 2.5cm height) for standardization.

Patties were cooked in an air-forced oven at either 180 °C or 220 °C. Cooking was terminated when core temperature of patties reached to 73 °C. After cooking, pork patties were covered by foil and chilled in cold room at 4 °C for 24 hours. Physical properties including texture and colour were measured on the following day of cooling. While part of samples were homogenized and stored in -18 °C for further chemical analysis. Cooking loss was determined according to the equation: Cooking loss (%) =  $(W_r - W_c)/W_r$ , where  $W_r$  was the weight of raw pork patties, and  $W_c$  was

weight of cooked pork patties. All the treatments were replicated three times. For each replicate, 8 patties were made for each treatment.

### **2.3 Composition analysis: pH, moisture, fat and protein content**

pH was measured by inserting a pH meter (68X243601, Oakton Instruments, USA) into the mixture by blending 5g meat sample with 45ml distilled water (Puangsombat et al., 2011). The moisture content was determined by drying 3g meat samples in an oven at 100°C for 24 hours according to AOAC methods (Horwitz & Latimer, 2005). The fat content was determined by drying sample in an oven for 4 hours firstly then using Soxhlet extraction system. The protein content was determined using the Kjeldahl method (Horwitz & Latimer, 2005).

### **2.4 Lipid/protein oxidation and antioxidant capacity**

#### **2.4.1 Lipid oxidation--Thiobarbituric acid-reactive substances (TBARS) value**

The degree of lipid oxidation in samples was expressed by TBARS values, which were determined by the method reported by Rodríguez-Carpena et al. (2012). 5 g well blended pork patty was homogenized with 15 ml perchloric acid (3.86%) and 0.5ml BHT (4.2% in ethanol) in a beaker, which was immersed in an ice bath to minimize oxidative reactions in samples during extraction. The mixture was then filtered and centrifuged at 3000 rpm for 4 min, 2 ml supernatant was mixed with 2 ml thiobarbituric acid (0.02 M) in test tube. The test tubes were then placed in a boiling water bath (100 °C) for 45 min. After cooling, the absorbance was measured at 532nm using a spectrophotometer (6315, Bibby Scientific Ltd, UK). The standard curve was prepared using 1,1,3,3-tetraethoxypropane (TEP) in 3.86% perchloric acid with the concentration of 0, 0.5, 1.0, 2.5, 5.0 and 10.0µM.

#### 2.4.2 Total protein carbonyl value (Protein oxidation)

The degree of protein oxidation can be evaluated by calculating the total carbonyl value according to the method described by Rodríguez-Carpena et al. (2012). 1g of well blended pork patties was homogenized at 1:10 (w/v) in 20 mM sodium phosphate buffer containing 0.6 M NaCl (pH 6.5) for 30 s. Two equal aliquots of 0.2ml mixture were then dispensed in 2 ml eppendorf tubes, respectively. 1ml cold trichloroacetic acid (TCA) (10%, w/w) was added into tubes and centrifuged for 5 min at 5000 rpm. One pellet was mixed with 1 ml 2 M HCl in order to measure protein concentration, while the other pellet was mixed with 1 ml of 0.2% (w/v) dinitrophenylhydrazine (DNPH) in 2 M HCl in order to measure carbonyl concentration. Both tubes were incubated for 1 h at room temperature. Subsequently, 1 ml 10% TCA was added into tubes and pellets were washed twice with 1 ml ethanol: ethyl acetate (1:1, v/v) to remove excess of DNPH. The pellets were then mixed with 1 ml of 20 mM sodium phosphate buffer containing 6 M guanidine HCl (pH 6.5), stirred and centrifuged for 2 min at 5000 rpm to remove insoluble fragments. Protein concentration was calculated from absorption at 280 nm using BSA as standard. The amount of carbonyls was expressed as nmol of carbonyl per mg of protein using an absorption coefficient of  $21.0 \text{ nM}^{-1} \text{ cm}^{-1}$  for absorbance at 370 nm for DNPH.

#### 2.4.3 Trolox equivalent antioxidant capacity (TEAC) of vegetable oil/ back fat

TEAC was used to evaluate the total antioxidant capacity of vegetable oils and pork back fat. The measuring procedures were based on the method reported by van den Berg et al. (1999). An ABTS radical solution was prepared by mixing 2.5 mM ABAP with 20 mM ABTS<sup>2-</sup> stock solution in 100 mM phosphate buffer (pH 7.4), which

contained 150 mM NaCl. The solution was covered with foil and heated at 60°C for 12 min, then cooled down to room temperature. 40 µl of the sample solution was mixed with 1960 µl of the freshly prepared ABTS/ABAP solution. Difference of absorbance at 734 nm in 6 min was recorded. A calibration curve was made by measuring the difference of absorbance in 6min for *Trolox* at the concentration of 0, 0.5, 1.0, 2.5, 5.0, 7.5 and 10.0 µM. The TEAC of vegetable oil or back fat was presented on a molar basis to *Trolox* (µmol Trolox/100g).

## 2.5 Analysis of HCAs

HCAs extraction and purification were following the procedures reported by Puangsombat et al. (2011). To minimize the variation and bias due to the unevenly distribution of sauce on the surface of meat, all samples were blended well with sauce before measuring. 3g ground meat sample was mixed with 12ml 1M sodium hydroxide firstly, then the mixture was transferred into an *Extrelut 20 column* with 17g diatomaceous earth. The HCAs were eluted by 60ml ethyl acetate in *Extrelut column*, and transferred into PRS cartridge which was pre-conditioned with 7ml ethyl acetate. A PRS cartridge was then washed with 6ml 0.1M HCl, 15ml methanol/0.1M HCl (45/55, v/v) and 2ml pure water to remove interferences from the PRS cartridge. The HCAs were then eluted by 20ml 0.5M ammonium acetate (pH 8.5) from the PRS cartridge and transferred into a C-18 cartridge that was conditioned with 5ml methanol and 5ml pure water. Finally, HCAs were eluted with 1ml methanol/ammonium hydroxide (9/1, v/v) from C-18 cartridge into 2ml vial, followed by drying the mixture under nitrogen stream for 1.5h at room temperature. The residue in the vial was dissolved with 1ml methanol and submitted for HPLC analysis. 1ml of mixtures containing 5 standard compounds (IQ, MeIQ, MeIQx, 4, 8-DiMeIQx

and PhIP) (5 ng/ml) was spiked into samples before extraction for measuring the recovery rate. Three replicates were carried out for each sample.

IQ, MeIQ, MeIQx, 4, 8-DiMeIQx and PhIP were analysed using HPLC (HP1635 Series, Agilent ChemStation, Agilent Technologies, Kidlington, UK) connected with a diode array UV detector (RF 2000). The HCAs were separated gradually by a reversed-phase Luna 5u C18 column (250 × 4.60 mm, 5 µm, 100A, Product No: 00G-4252-E0, Phenomenex, UK). Mobile phase A was 0.01 M triethylamine (adjusted pH 3.6 with phosphoric acid) and phase B was acetonitrile (>99%, HPLC grade). The solvent contained 95% A and 5% B at beginning, then linearly changed to 75% A and 25% B within 30 min at flow rate 1.0 ml/min. The temperature of column was 40°C. The UV detector was set at 252 nm (Puangsombat et al., 2011).

## 2.6 Analysis of PAHs

5g meat sample was homogenised with 15ml 1M NaOH for 1 h. The homogenized sample was then mixed with 17g diatomaceous earth and loaded in an *Extrelut 20 column*. Elution of PAHs started from *Extrelut column*, and was followed by propyl sulphonic acid (PRS) cartridges with 60 ml of CH<sub>2</sub>Cl<sub>2</sub> containing 5% toluene. The dichloromethane solution was then evaporated to small amount (0.5-1ml) and the rest of the solvent was dried under a nitrogen stream. The residue was re-dissolved in 1ml n-hexane and transferred to the top of a glass column packed with silica-gel (10 g). PAHs were then eluted by 25 ml of n-hexane and 60 ml of 60:40 (v/v) n-hexane–CH<sub>2</sub>Cl<sub>2</sub> mixtures. After evaporation to dryness the residues were dissolved in acetonitrile (spiked and unspiked samples) before the HPLC analysis (Janoszka, 2011). 50ng of 2 standard mixtures (BaA and BaP) was spiked for measuring the recovery rate.

BaA and BaP were analysed using HPLC (HP1635 Series, Agilent ChemStation, Agilent Technologies, Kidlington, UK) connected with a fluorescence detector. Mixture of 84% acetonitrile (>99%, HPLC grade) and 16% water (HPLC grade) were used as a mobile phase under isocratic conditions. The separations were performed at 40°C under isocratic conditions with flow rate 1.0 ml/min. The fluorescence detection was performed by applying the following excitation (Ex)/emission (Em) wavelength program: 280/410 nm from 0 to 8.50 min (BaA), 376/410 nm from 8.50 to 15 min (BaP) (Janoszka, 2011).

## 2.7 Recovery rate of HCAs and PAHs

The 5 standard HCA (IQ, MeIQ, MeIQx, 4, 8-DiMeIQx and PhIP) and 2 PAHs (BaA and BaP) compounds were identified through the retention time of the peaks, and the quantity of each individual compound was determined according to the standard calibration curves, which was established by the standard solution at 0.5ng/ml, 5ng/ml and 50ng/ml. Limit of detection (LOD) for IQ, MeIQ, MeIQx, 4, 8-DiMeIQx, PhIP, BaA and BaP were for 0.02 ng/g, 0.01 ng/g, 0.02 ng/g, 0.05 ng/g, 0.03 ng/g, 0.07 ng/g and 0.06 ng/g. The average recoveries of these 5 HCAs based on triplicates were 60.01% for IQ, 61.76% for MeIQ, 53.64% for MeIQx, 60.57% for 4,8-DiMeIQx and 55.98% for PhIP. Results were comparable with several published data (Gibis, Kruwinnus & Weiss, 2015; Messner & Murkovic, 2004; Oz & Cakmak, 2016; Yao et al., 2013). The recovery rate for BaA and BaP was 54.37 and 49.54% respectively, which was comparable with published results of 50% - 115% (Farhadian, Jinap, Abas, & Sakar, 2010; Ishizaki, Saito, Hanioka, Narimatsu, & Kataoka, 2010; Iwasaki et al., 2010; Janoszka, 2011). Recovery rate could be affected by sorbing materials, flow rate through cartridges, organic modifier quality and/or content, interfering effects of eluting solvents (Buseti, Heitz, Cuomo, Badoer,

& Traverso, 2006). Stevens, Hamstra, Hegeman & Scharrer (2006) reported that overlapping peaks on chromatogram caused by insufficient separation procedures could result in recovery rate over 110%, and AOAC (2016) recommends that 40-120% recovery rate is acceptable for compounds at 1ng/g concentration.

## 2.8 Inhibitory rate of HCAs and PAHs

Inhibitory rate was determined according to the equation:

$$\text{Inhibitory rate (\%)} = (A_c - A_t) / A_c \times 100$$

where  $A_c$  was the total amount of HCAs/PAHs in control samples (ng/g), and  $A_t$  was the total amount of HCAs/PAHs in fat partially replaced patties (S/ O/ G patties) (ng/g).

## 2.9 Physical parameters

### 2.9.1 Colour

Colour feature including  $L^*$ ,  $a^*$  and  $b^*$  was measured using Hunter Lab Colour instrument (Hunter Associates Laboratory, Virginia, USA, 2003). Each sample was measured at 3 different locations, while the average was recorded as the colour feature of the sample.

### 2.9.2 Texture

Texture profile analysis (TPA) was performed at room temperature with a *Stable system* Texture Analyzer (Middleboro, USA). Cylinder samples (D=18mm, H=22mm) were prepared using a stainless cork borer. Before the analysis, samples were tempered at room temperature (20°C) for 30min. The settings for texture analysis were: load cell 5 kg, head speed 1.6 mm/s, and compression depth 10.0 mm. The



calculation of TPA values was based on the compression curve with force (y-axis) and time (x-axis). Values for hardness (N) was defined as the absolute peak force in the 1<sup>st</sup> compression cycle, cohesiveness as the area of work in the 2<sup>nd</sup> compression divided by the area of work in the 1<sup>st</sup> compression, chewiness as the product of hardness\*cohesiveness\*springiness, and springiness as the force that sample return to its initial pattern after compression (Sánchez del Pulgar, Gázquez & Ruiz-Carrascal, 2012). Each sample was measured at least 5 times, and the average was recorded as the value of the sample.

## 2.10 Statistical analysis

Each treatment had three replicates. Statistical significance test was carried out by using SPSS Statistics 21 (IBM, 2014). The significant difference in chemical composition, physical property, levels of HCAs and PAHs for the 8 treatments were carried out by one-way analysis of variance (ANOVA) at the significant level 0.05, and *Duncan* test was selected for multiple comparison if equal variances assumed, otherwise *Tamhane's T2* test was used. To analyse the effect of factors and the interaction between factors (cooking temperature and replacing oil type), two-way ANOVA was employed at the significant level 0.05. Multivariate linear regression model was employed to explore the effect of multi independent factors, including lipid oxidation, protein oxidation and antioxidant capacity of oils on the formation of HCAs and PAHs at the significant level 0.05. Pearson correlation was employed for the measurements as a prerequisite for the regression.

## 3. Results and discussion

### 3.1 Effect of replacing oil and cooking temperature on proximate composition and physical parameters of reduced fat pork patties

### 3.1.1 Proximate composition

Table 2 shows that the effect of replacing oil and cooking temperature on proximate composition and pH values in patties. Type of replacing oil did not pose any effect on all proximate composition and pH, but cooking temperature had significant influence on moisture and cooking loss ( $p<0.01$ ). Interaction between type of oil and cooking temperature was only observed in cooking loss ( $p<0.01$ ).

Moisture content varies from  $63.36 \pm 0.37\%$  to  $67.53 \pm 0.26\%$  in fat partially replaced patties, which are consistent with the results reported by Rodríguez-Carpena et al. (2012). They found that pork patties with 50% fat replaced with sunflower oil, olive oil and avocado oil had moisture level at 61.48-63.39% when patties were cooking at 170 °C for 18 min in a forced-air oven. There was no significant difference in moisture content for all the patties (with or without fat replacement) ( $p=0.206$ ), which indicated that replacing back fat with vegetable oils did not affect moisture level in the final products. However, cooking temperature significantly affected the moisture content in the final products ( $p<0.05$ ). Higher cooking temperature at 220 °C led to low moisture level in cooked patties, compared with low cooking temperature at 180 °C ( $p<0.01$ ), which were 63.58% vs 69.15% for control, 63.36% vs 65.91% for olive oil treatment, 63.49% vs 66.34% for sunflower oil treatment and 65.90% vs 67.53% for grape seed oil treatment. Low moisture level in the final products are directly associated with high cooking loss, as cooking loss is mainly composed of water and water soluble nutrients such as proteins (Sánchez del Pulgar et al., 2012). Pork patties with or without fat replacement had cooking loss ranging from 20.30%- 24.75%, which was consistent with the results reported by Rodríguez-Carpena, Morcuende and Estévez (2011). They found that cooked pork

patties at 170 °C for 18 min had cooking loss at 20.69%- 22.20%. Fat and protein content ranged from 9.49%-10.01% and 15.03%-15.34% respectively, as expected, fat and protein content in fat modified patties were comparable with these in C patties. pH ranged from 5.85 to 5.99. Cooking temperature and type of oil did not affect pH of patties, there was no interaction between temperature and type of oil as well ( $p>0.05$ ).

### 3.1.2 Physical parameters

#### 3.1.2.1 Texture

Texture property of cooked pork patties was examined including hardness, cohesiveness, springiness and chewiness through a typical texture profile analysis. Results were listed in Table 3. Cooking temperature affected hardness, cohesiveness and chewiness ( $p \leq 0.05$ ), but had no effect on springiness ( $p>0.05$ ). Fat replacement with vegetable oils did not affect any of the texture attributes ( $p>0.05$ ). There was no interaction observed between oil replacement and temperature for hardness, chewiness and springiness except cohesiveness ( $p>0.05$ ). High cooking temperature led to high hardness. Control patties cooked at 220°C had significantly higher hardness ( $26.65 \pm 3.15\text{N}$ ) than that cooked at 180°C ( $20.14 \pm 2.81\text{N}$ ) ( $p<0.05$ ). Roldán et al. (2013) observed that the elevating cooking temperature resulted in higher hardness, while the increased hardness in pork patties might be associated with high cooking loss. There was no difference observed in hardness for fat modified patties regardless of temperature ( $p>0.05$ ), which agreed with the results reported by Rodríguez-Carpena et al. (2011). They stated that patties that partially (50%) replaced with sunflower oil, olive oil and avocado oil had same hardness with control patties. However, Hur, Jin and Kim (2008) reported that olive oil replacement

in pork patties resulted in low hardness compared with control patties. The cooking temperature could help to explain the disagreement. In their study, patties were cooked at 100 °C in water bath, while samples were cooked in convection oven at 180°C or 220°C. The hardening effect of high cooking temperature could be neutralized by the softening effect of to the replacing vegetable oil. As a result, there was no difference observed in hardness for fat modified patties and control sample ( $p>0.05$ ).

Temperature affected cohesiveness greatly ( $p=0.001$ ), because the texture parameters are mainly determined by denaturation of the structural protein system, i.e. actomyosin complex and collagen (Palka & Daun, 1999). Type of oil did not have any effect on the cohesiveness of pork patties ( $p>0.05$ ). Rodríguez-Carpena et al. (2012) also reported that there was no difference in cohesiveness between control and patties with 50% fat substitution using avocado, sunflower or olive oil. The interaction between temperature and type of oil on cohesiveness was observed ( $p<0.05$ ). For control, olive and sunflower oil treatment, patties cooked at 220°C had higher cohesiveness than these cooked at 180°C ( $p<0.05$ ), but there was no difference in cohesiveness for grape seed oil samples cooked at different temperatures ( $p>0.05$ ). This interaction between cooking temperature and type of oil could be explained by the emulsion stability of oil/fat emulsion. Youssef and Barbut (2009) and Rodríguez-Carpena et al. (2011) reported that vegetable oil with high polyunsaturated fatty acids (PUFA) had small fat globules in meat emulsion, which could offer a stronger fat-protein interaction. Grape seed oil contained high level of PUFA compared with back fat and olive oil, which led to a stable emulsion in G patties. In addition, polyphenol compounds in grape seed oil emulsion could help maintain the protein functionality through inhibition of protein oxidation during

processing (Ganhão, Morcuende & Estévez, 2010). Thus, grape seed oil/meat emulsion was less sensitive to temperature changes in relation to cohesiveness of the final products.

For chewiness, temperature significantly affected it ( $p < 0.05$ ). High cooking temperature resulted in high chewiness. Chewiness remained similar for all patties cooked at 180 °C (from 3.94 to 4.11 N.s), but increased to  $5.92 \pm 0.77$  N.s in C patties, and  $5.35 \pm 0.41$  N.s in S patties that cooked at 220 °C. Greater hardness, cohesiveness and chewiness at higher elevating cooking temperature could be due to the more severe denaturation of myosin (40-60 °C) and actin (66-73 °C) (Sánchez del Pulgar et al., 2012). In addition, chewiness is also associated with the water retention in meat products (Roldán et al., 2014). Patties cooked at 180°C had significantly higher moisture content than those cooked at 220°C (as shown in Table 2). Therefore, high chewiness would be expected in samples cooked at 220°C due to high moisture loss.

### 3.1.2.2 Colour

Effects of vegetable oil and cooking temperature on colour characteristics of cooked pork patties including lightness ( $L^*$ ), redness ( $a^*$ ) and yellowness ( $b^*$ ) were summarized in Table 3. Temperature significantly affected all three parameters ( $p \leq 0.05$ ), especially  $a^*$  with  $p = 0.005$ . Different vegetable oils did not have any impact on the colour parameters, while the interaction between oil and temperature was observed in yellowness ( $b^*$ ). Patties cooked at 220 °C had lower  $L^*$  than those cooked at 180 °C ( $p < 0.01$ ), which agreed with the results of Sánchez del Pulgar et al. (2012). The decrease in surface lightness could be attributed to the brown pigments formed from caramelization of sugars and Maillard reaction when samples were

cooked at temperature over 90 °C (Girard, 1992). In addition, the lightness was also associated with the moisture content in meat products. Qiao et al. (2001) reported that there was positive correlation between lightness and moisture content in broiler breast fillet. Presence of heme pigments, containing 90-95% myoglobin in muscles gives meat red colour ( $a^*$ ). At 180 °C, a higher  $a^*$  value was found in all pork patties with oil replacement than control patties ( $p < 0.05$ ). The antioxidants in these vegetable oils, such as vitamin E could prohibit the oxidation of oxymyoglobin and lead to a high redness in the final products (Hui, 2001; Sánchez del Pulgar et al., 2012). Cooking temperature could significantly affect  $a^*$  as well. All patties that cooked at 220 °C had significantly lower  $a^*$  than those cooked at 180 °C. The reduction of  $a^*$  caused by increased temperature could be associated with the denaturation of myoglobin (Nollet, 2012). Liao, Xu and Zhou (2009) found that  $a^*$  of stir fried pork floss decreased significantly by 30% when cooking temperature increased from 125 °C to 150 °C.

Yellowness  $b^*$  ranged from 15.41-18.73 in all cooked patties. Both type of oil and cooking temperature had no effect on  $b^*$  values, but the interaction between type of oil and cooking temperature was observed. Jamali et al. (2016) also found that  $b^*$  value in beef patties was not affected by cooking temperature (160 °C and 220 °C). The results of  $b^*$  in control samples (16.98-18.73) were comparable with Vittadini et al. (2005).

### **3.2 Effects of vegetable oils and cooking temperature on the formation of HCAs**

Concentration of HCAs (IQ, MeIQ, MeIQx, 4, 8-DiMeIQx and PhIP) in control patties and fat modified patties cooked at 180°C and 220°C were listed in Table 4. Type of

oil affected all individual HCAs compound except IQ, cooking temperature significantly affected the total amount of HCAs and all individual HCAs compounds except MelQx. Interaction between oil and cooking temperature was observed in total HCAs, IQ, MeIQ, 4, 8-DiMeIQx and PhIP, but not MelQx. At both temperatures, all fat modified patties had significantly lower amount of MeIQ, 4, 8-DiMeIQx and total HCAs than control patties ( $p < 0.05$ ). MelQx, 4, 8-DiMeIQx and PhIP were not detectable in G patties. Tocopherols (average 50mg/100g in sunflower oil, 100-300mg/g in refined olive oil and 10-15mg/g in grape seed oil) and polyphenols (common profile in grape seed oil: catechin, epicatechin and epicatechin gallate) in these oils could play roles in reducing the final HCAs in patties (Bail et al., 2008; McCance & Widdowson, 2002; Rombaut et al., 2014). Tocopherols have been found to block dialkyl-pyrazine radicals for further reaction with creatine to form HCAs, or react with precursors of 4, 8-DiMeIQx to inhibit the formation of HCAs (Pearson et al., 1992; Vitaglione & Fogliano, 2004). Phenolic compounds could also prevent the formation of imidazoquinoline-type HCAs through trapping pyrazine cation radicals and some other carbon-centred radicals generated either from pyrazine cation radicals or different pathway during Maillard reaction (Kato et al., 1996). In addition, phenolic compounds have the ability to directly trap phenylacetaldehyde, which is a major intermediate during the formation of PhIP (Cheng et al., 2007).

Total HCAs ranged from not detected (Nd) to  $140.57 \pm 22.03$  ng/g. Control patties cooked at both cooking temperatures contained significantly higher amount of total HCAs ( $67.56 \pm 17.29$  ng/g and  $140.57 \pm 22.03$  ng/g), followed by S patties ( $5.98 \pm 1.10$  ng/g and  $23.88 \pm 2.44$  ng/g) and O patties ( $4.11 \pm 0.87$  ng/g and  $20.03 \pm 2.25$  ng/g), while G patties achieved the lowest total HCAs in both temperatures (Nd and  $1.90 \pm 0.04$  ng/g). Control samples cooked at 220°C contained all types of HCAs,

whereas none of HCA compounds were detected in G patties cooked at 180°C. The dominating compounds of HCAs were MeIQ (59.70 ± 0.98ng/g) and 4, 8-DiMeIQx (43.37 ± 15.67ng/g) in C patties, while PhIP in S (14.78 ± 1.49ng/g) and O patties (22.70 ± 1.95ng/g). The total HCAs in C patties were higher than some published results. Vangnai et al. (2014) reported MeIQx (7.59 ± 0.43ng/g), PhIP (13.12 ± 0.72 ng/g) and total HCAs (22.35 ± 1.17 ng/g) in fried pork loins cooked at 204 °C for 16 minutes. The total level of HCAs in pan-fried well-done pork was 49.7ng/g with cooking ended at 80 °C core temperature (Iwasaki et al., 2010). The sampling procedure for measuring HCAs could help explain the difference. HCAs were extracted from the 2mm outer layer surface of samples in this study, while lots of researchers extracted HCAs from entirely ground samples. The precursors of HCAs, such as creatine, glucose and free amino acids would migrate to the surface of meat and enhance Maillard reactions during cooking (Gibis & Weiss, 2015). As a result, surface could accumulate much higher level of HCAs compared with internal part of the sample. Therefore, a higher concentration of HCAs would be expected than that exacted from entirely ground samples.

IQ was detected up to 3.88 ng/g in cooked patties, which was in the range of 0.7-5.3 ng/g in fried ground beef patties (Balogh et al., 2000). At 180 °C, IQ was not detected apart from O patties, but cooking at 220°C generated high level of IQ in all patties (p<0.05). Different vegetable oils did not affect the formation of IQ (p>0.05), but interaction between vegetable oil and temperature was observed in formation of IQ. IQ is generally formed through reactions between creatinine, pyridine radicals and formaldehydes (Vitaglione & Fogliano, 2004). Vegetable oils could decompose into hydroperoxides, and then aldehydes and ketones at high cooking temperature, which further react with amino acids in Maillard reactions (Johansson, Skog, &



Jagerstad, 1993; Zamora & Hidalgo, 2007). Olive oils with high level of oleic acid could decompose into aldehydes (-CHO) much faster (3-15 times) than sunflower oil and grape seed oil containing high amount of linoleic and linolenic acid at 190°C (Guillén & Uriarte, 2012b). This might explain why IQ was only detected in olive oil patties at 180°C. But the pathway of formation of HCAs in the real meat system was more complicated than chemical model system, and antioxidants in oils should also be considered (Johansson, Skog, & Jagerstad, 1993). Therefore, further investigation is needed to understand promoting mechanism and role of antioxidants in oils. The highest MeIQ was found in control patties cooked at 220 °C ( $59.70 \pm 0.98$  ng/g), followed by in control patties cooked at 180 °C ( $18.26 \pm 14.46$  ng/g). Janoszka (2010) reported that 6.28 ng/g MeIQ was detected in pan-fried pork patties cooked at 170°C for 12 minutes, which was similar with this study. Formation of MeIQ was completely inhibited by olive oil and sunflower oil at both cooking temperatures. Grape seed oil could only inhibit formation of MeIQ at low cooking temperature, while 1.31ng/g was detected in G patties cooked at 220°C. The inhibitory effect on MeIQ could be attributed to the antioxidants such as vitamin E and polyphenols in vegetable oils. Rounds et al. (2012) and Liao et al. (2009) also reported that vitamin E and polyphenols could reduce the formation of MeIQ. Balogh et al. (2000) found that vitamin E had stronger inhibitory effect on MeIQ with reduction rate 64.3% than phenolic compound in Oleoresin rosemary extract with reduction rate 47.9% in fried beef patties. Since olive oil and sunflower oil contain higher level of vitamin E than grape seed oil, therefore, stronger inhibition of MeIQ would be expected in O patties and S patties. Cooking temperature did not affect MeIQx level in patties ( $p > 0.05$ ), but formation of MeIQx was significantly affected by different vegetable oils. S patties and O patties had similar MeIQx with control sample. For G patties cooked at both

temperatures, there was no MeIQx detected. However, sunflower oil and olive oil did not affect MeIQx in patties, although both oils were rich of vitamin E. No MeIQx was detected at all which was in agreement with Rounds et al. (2012), who also reported grape seed extract could completely inhibit the formation of MeIQx in cooked beef patties. Temperature significantly increased formation of 4, 8-DiMeIQx as evidenced in C patties cooked at 180 °C and 220 °C ( $25.66 \pm 1.51$  ng/g and  $43.37 \pm 15.67$  ng/g) ( $p < 0.05$ ). All vegetable oils effectively reduced 4, 8-DiMeIQx in patties. Grape seed oil was the most effective one among the three vegetable oil as 4, 8-DiMeIQx was not detected in G patties cooked at both temperatures.

PhIP ranged from Nd to  $24.07 \pm 1.99$  ng/g. A similar level of PhIP ( $18.4 \pm 11.5$  ng/g) in the fried pork patties was reported by Zhang et al. (2013), when patties were cooked at 180°C for 5 min. At 180 °C, PhIP ( $11.43 \pm 6.33$  ng/g) was totally inhibited by all 3 vegetable oils, but only grape seed oil could completely inhibit the formation of PhIP at both cooking temperatures. The stronger inhibitory effect on PhIP in G patties could be attributed to the phenolic compounds in grape seed oil. Gibis and Weiss (2012), Jamali et al. (2016) and Oguri et al. (1998) found that catechin, epicatechin and epicatechin-3-O-gallate in grape seed extract may be responsible for 50%-90% reduction of PhIP in both oven cooked beef patties and chemical model system. Zamora and Hidalgo (2015) suggested phenolic compounds could effectively scavenge the carbonyl compounds in the Strecker degradation of phenylalanine to produce phenylacetaldehyde (major intermediate in the development of PhIP). Temperature significantly affected PhIP level in pork patties ( $p < 0.05$ ). PhIP increased significantly in C patties from  $11.43 \pm 6.33$  ng/g to  $24.07 \pm 1.99$  ng/g, O patties from Nd to  $14.78 \pm 1.49$  ng/g and S patties from Nd to  $22.70 \pm 1.95$  ng/g when cooking temperature increased from 180 °C to 220 °C ( $p < 0.05$ ). The results agreed

with Gibis and Weiss (2012) and Wong et al. (2012) that PhIP level was directly related to the cooking temperature. At 175-200 °C, only very low level of PhIP (0-6.91 ng/g) could be formed even at varied cooking time, but it could increase dramatically to 31.80ng/g with prolonged cooking time if temperature went above 200 °C.

Type of vegetable oil significantly affected the level of total HCAs in cooked patties ( $p<0.05$ ), but temperature did not have any effect ( $p>0.05$ ). Interaction between types of vegetable oil and cooking temperature was observed as well ( $p<0.05$ ). In the current study, reduction of total HCAs by 85.75%-93.90% was found in O patties, 83.01%-91.15% in S patties, while G patties achieved the highest reduction rate at 98.64%-100% (Table 4). Antioxidants in the vegetable oils could be responsible for the reduction of total HCAs as a strong negative correlation ( $r=-0.618$ ,  $p<0.01$ ) was disclosed between total HCAs level and antioxidant capacity (TEAC) of oils (Table 6). Grape seed oil had the highest TEAC value with  $0.71\pm0.01$   $\mu\text{mol Trolox}/100\text{g}$ , followed by olive oil ( $0.52\pm0.05$   $\mu\text{mol Trolox}/100\text{g}$ ) and sunflower oil ( $0.18\pm0.04$   $\mu\text{mol Trolox}/100\text{g}$ ), while pork back fat had the lowest TEAC value ( $0.09\pm0.02$   $\mu\text{mol Trolox}/100\text{g}$ ) (Figure 1). Therefore, reduction of HCAs in vegetable oils, especially grape seeds oil was expected compared with control sample. Results were comparable with findings of Matthäus (2008) and Castelo-Branco and Torres (2012). Balogh et al. (2000) found that 1% vitamin E spray on the surface of beef patties could reduce the concentrations of IQ, MeIQ, MeIQx, DiMeIQx and PhIP significantly by 45% to 75%, because vitamin E could remove free radicals in Maillard reactions. Similar result was also reported by Lan, Kao and Che (2004). They found that 70% of total HCAs (IQ, MeIQ, MeIQx, 4, 8-DiMeIQx and PhIP) were prohibited when 0.2%  $\alpha$ -tocopherol was added into ground pork 1h before cooking. Phenolic compounds,

such as catechin, epicatechin-3-O-gallate, oligomer procyanidins and tocopherols in grape seed oil contributed to its antioxidant capacity (Agostini et al., 2012; Crews et al., 2006; Matthäus, 2008). Vitaglione and Fogliano (2004) suggested that mixture of antioxidant compounds could perform better than single antioxidant as they could inhibit various pathways in different steps of reactions. Therefore, phenolic compounds might work synergistically with tocopherols to enhance each other to inhibit the formation of HCAs. However, the synergistic effect between different antioxidants needs to be further examined.

Cooking temperature significantly affected total HCAs in pork patties ( $p < 0.01$ ) (Table 4). Patties cooked at 220°C had significantly higher level of total HCAs than those at 180°C ( $p < 0.01$ ). Effect of temperature on the formation of HCAs was well examined in previous research (Knize et al., 1994; Liao et al., 2009; Oz & Kaya, 2011). Thermal processing has vital influence on the formation of polar HCAs (IQ, MeIQ, MeIQx, DiMeIQx and PhIP), which are formed in meat products when samples are cooked at 160-250 °C, typical domestically cooking temperature. High cooking temperature generated more diverse types of HCAs, but also stimulate the accumulation of the amount of HCAs on the surface of meat products (Olsson & Pickova, 2005; Skog, Johansson & Jaegerstad, 1998).

### **3.3 Effects of vegetable oils and cooking temperature on the formation of PAHs**

Concentration of PAHs (BaA and BaP) in cooked pork patties with different cooking temperature was listed in Table 5. The range of total PAHs was from  $1.59 \pm 0.26$  ng/g to  $3.84 \pm 0.21$  ng/g. The dominating compound of PAHs was BaP in all samples. BaA ranged from 0.14-0.31 ng/g in cooked patties, while BaP ranged from 1.44 to

3.53ng/g. Temperature did not affect the formation of both BaA and BaP, but type of vegetable oil had significant effect on the formation ( $p<0.05$ ). Interaction between type of oil and cooking temperature was also observed in both compounds (Table 5). BaP level in this study (1.44-3.53 ng/g) are consistent with results reported by Nisha et al. (2015) and Janoszka (2011), i.e. 1.52 ng/g of BaP in the oven broiled pork and 1.61 ng/g BaP in oven grilled pork chop (17min at 170°C). Olive oil and grape seed oil showed inhibitory effect on BaP when patties cooked at 220 °C, but no effect or even promoting effect was observed at 180 °C. On the contrast, sunflower oil offered inhibition on BaP at 180 °C, but promotion at 220 °C. As BaP is one of the highest toxic potency compounds during meat cooking, EU Commission has regulated that the updated limit of BaP occurring in processed meat and seafood products is 2 ng/g (Purcaro et al., 2013; Wretling et al., 2010). Among all the patties, only O patties cooked at 220 °C and S patties cooked at 180 °C met the safety regulation of BaP. Therefore, it is necessary to develop any procedures or alternative methods that reduce the amount of BaP to safety limit.

The effect of oil and interaction between type of oil and cooking temperature on the formation of PAHs were significant ( $p<0.05$ ). Cooking temperature did not affect the formation of PAHs ( $p>0.05$ ). S patties cooked at 220 °C had the highest total PAHs ( $3.84\pm0.21$  ng/g), followed by G patties ( $3.46\pm0.16$  ng/g) cooked at 180 °C and C patties ( $3.28 \pm 0.07$  ng/g) cooked at 220 °C. O patties cooked at 220°C obtained the lowest PAHs. PAHs were mainly associated with the pyrolysis of fat undertaken at high temperature (Viegas et al., 2012). Therefore, smoking point of vegetable oils may help explain the difference in PAHs. Sunflower oil and grape seed oil contain high content of PUFAs, especially linolenic acid and linolenic acid have lower smoke points (grape seed oil 216°C, sunflower oil 227°C) which makes them easy to

decompose, compared with olive oil (smoke point 242°C). The decomposition of oil could generate more reactive free radicals to accelerate the production of PAHs (Chen & Lin, 1997; Elmore et al., 2002). They also concluded that hydroperoxides from lipid oxidation, could subsequently generate cyclic compounds through intramolecular reaction, and PUFA could undergo further polymerization. In addition, vegetable oils themselves contained BaP, which might increase the total amount of PAHs in cooked meat. Fromberg, Højgard and Duedahl-olesen (2007) reported that olive oil, sunflower oil, and grape seed oil approximately contained 0.12 ng/g, 0.4 ng/g and 1.0 ng/g BaP respectively. As a result, high level of PAHs was expected in sunflower oil and grape seed oil samples. Although vegetable oils contain antioxidants, the inhibitory effect on PAHs formation was not observed consistently. In S patties, the inhibitory efficiency at both temperature were 7.75% and 51.52%, but olive oil and grape seed oil increased the formation of PAHs by 17.07% and 34.11%, respectively. It shows that antioxidants in vegetable oils were not involved in the formation of PAHs to a great extent, which is further confirmed by the correlation analysis. As indicated in Table 6, there is no correlation relationship observed between antioxidant capacity of oil (TEAC) and total PAHs. The impact of tocopherols and phenolic compounds on the formation of PAHs in processed meat were not well documented. In vitro study, Zhu et al. (2014) found that vitamin E intake could significantly prohibit free radicle induced by BaP and protect cellular damage in human lung, but the effect antioxidants on formation of PAHs in food products has been scarce.

#### **3.4 Correlation between lipid oxidation, protein oxidation and the formation of HCAs and PAHs**

In Figure 2, S patties, O patties and G patties had significantly lower TBARS values than control samples ( $p < 0.05$ ), while G patties had the lowest TBARS value compared with O and S patties ( $p = 0.001$ ). This inhibitory effect against lipid oxidation could be attributed to the antioxidants (tocopherols and phenolic compounds) within the oils, since a significantly negative relationship was found between TBARS values and antioxidant capacity of oils ( $r = -0.764$ ,  $p < 0.01$ ). Wong et al. (2015) reported that 0.1-0.4 mmol vitamin E could inhibit 45% of lipid oxidation in beef patties, by obstructing the formation of some key aldehydes and ketones during lipid oxidation. Similar results were achieved by Ahn, Grün and Fernando (2002) as well. Frankel (1998) proposed that  $\alpha$ -tocopherol could prevent the chain propagating and remove free radicals through reacting with either singlet oxygen or peroxy radicals. Consequently lipid oxidation was reduced. Meanwhile, polyphenols, such as epicatechin (EC) oligomer procyanidins were also sufficient to inhibit lipid oxidation by reducing free radicals and preventing chain propagation in cooked pork and beef (Rojas & Brewer, 2007). They could also chelate metals (iron and copper in meat) or react with ROS, and then turn into non-radical species. As a result, reactions were terminated (Roman et al., 2013). Moreover, Ahn et al. (2002) and Tang et al. (2001) reported that polyphenols such as catechin, epicatechin were more efficient in inhibiting lipid oxidation than  $\alpha$ -tocopherol at the same concentration in cooked meat. This could explain why G patties had the lowest TBARS values than S and O patties. Although Gunstone (2002) stated that a higher degree of unsaturation of fatty acids could be easier to trigger the lipid oxidation and interacted with Maillard reaction, the presence of antioxidants should be also considered in the formation of HCAs.

Protein carbonyls are produced from protein oxidative degradation in meat products, which were used to analyse degree of protein oxidation (Figure 3). Significant effect

of vegetable oils on the protein oxidation was observed ( $p=0.001$ ). C patties had a significantly higher level of protein carbonyls (12.11 mmol/kg) than other 3 fat modified patties ( $p<0.05$ ). The protein oxidation could be inhibited by the presence of antioxidants in oil, as negative correlation between antioxidant capacity of oils (TEAC) and protein carbonyl level was found with  $r=-0.606$ ,  $p<0.01$ , as indicated in Table 6. Botsoglou et al. (2014) found that protein carbonyl value could be reduced from 3.25 mmol/kg to 2.25 mmol/kg in cooked pork patties when 50mg/kg  $\alpha$ -tocopherol was added. Vuorela et al. (2005) reported that phenolic compounds, including vinylsyringol and sinapic acid in rapeseed oil had good antioxidant capacity against protein oxidation in cooked pork patties. Ganhão et al. (2010) also found that arbutus-berries extract containing catechins significantly reduced protein oxidation by chelating heme iron in cooked patties. However, there was no difference in the protein carbonyl level among O, S and G patties ( $p>0.05$ ).

In the development of HCAs under high temperature cooking, both lipid oxidation and protein oxidation are involved with Maillard reactions in the meat system (Johansson, Skog & Jagerstad, 1993; Zamora & Hidalgo, 2007). Lot of researchers believed that formation of HCAs could be primarily related to interactions between free radicals generated from lipid oxidation and free radicals produced in Maillard reactions (Hwang & Ngadi, 2002; Skog et al., 1998). Therefore, it is useful to explore the relationship between lipid/protein oxidation and the formation of HCAs. In this study, correlation analysis was conducted between TBARS/protein carbonyl values and concentration of total HCAs in fat modified cooked patties (Table 6). Significant positive correlation was disclosed between total HCAs and TBARS ( $r=0.826$ ,  $p<0.01$ ) and between HCAs and protein carbonyl ( $r=0.778$ ,  $p<0.01$ ), which further confirmed that both lipid oxidation and protein oxidation participated the formation of HCAs



during cooking process. In order to further examine the relationship between lipid oxidation/protein oxidation/antioxidant capacity of lipids and total HCAs in cooked patties, multivariate linear regression model was displayed below,

$$\text{Total HCAs} = -42.37 + 108.26 * \text{TBARS} + 5.647 * \text{Protein Carbonyl}.$$

It can be seen from the equation that TBARS (lipid oxidation) played a predominant role on the formation of HCAs, compared with protein carbonyl (protein oxidation). The factor 'TEAC (antioxidant capacity of lipids)' has been removed from the model, because the strong correlation between TEAC and TBARS/protein carbonyl indicates that variance accounted for TEAC could be well accounted by TBARS/protein carbonyl. In cross-validation, adding/removing 'TEAC' caused little change in adjusted R square of the predicted models, which indicated variance caused by TEAC could be well explained by other independent factors in the model (Field, 2009).

Free radicals, such as aldehydes and ketones generated from lipids oxidation could interact with Maillard reactions by reacting with the polar head of an amino group to produce more HCAs (Jägerstad et al., 1998; Zamora & Hidalgo, 2007). On the other hand, active protein carbonyl residues, such as alkyl, peroxy radicals that formed by muscle protein oxidation can be initialized by lipid oxidation, metal ions and other peroxidized compounds (Cai et al., 2002). Subsequently, these carbonyls could interact with Maillard reaction via Schiff base and then generate Strecker aldehydes, which are intermediates of imidazoquinolines and imidazoquinoxalines (Estévez, 2011; Soladoye et al., 2015). Researchers also suggested that lipid oxidation could trigger protein oxidation by reacting with heme iron that released from myoglobin (Ganhão et al., 2010; Vuorela et al., 2005). In this work, there was no correlation

observed between TBARS /protein carbonyl and PAHs ( $p>0.05$ ), which indicated that the involvement of lipid and protein oxidation in the formation of PAHs was only at null level. Thus, no linear regression model was fitted. The antioxidants in vegetable oils could not inhibit the formation of PAHs, which was evidenced by null correlation between TEAC and PAHs ( $p>0.05$ ).

#### 4. Conclusions

Control patties contained the highest amount of HCAs and relatively higher PAHs at 180 °C and 220 °C. All 3 fat modified patties had significantly lower HCAs, which could be attributed to antioxidants, such as tocopherols and polyphenol compounds existing in the oils. The negative correlation ( $r= -0.618$ ,  $p<0.01$ ) between the antioxidant capacity of lipids and the total amount of HCAs could be useful evidence to support this claim. Both lipid and protein oxidation contributed to the formation of HCAs, which were supported by the positive relationship between TBARS/ protein carbonyl values and total HCAs with  $r= 0.826$  and  $0.788$  ( $p<0.01$ ), respectively. Olive oil and sunflower oil completely prohibited MeIQ, whereas grape seed oil could inhibit MeIQx, 4, 8-DiMeIQx and PhIP. Grape seed oil could achieve the highest inhibitive effect on the formation of HCAs. However, effect of vegetable oils on the formation of PAHs was not consistent, which could be attributed to complexity of oil decomposition and antioxidants in the oils. The involvement of lipid oxidation and protein oxidation in formation of PAHs was limited or at a minimum level. Antioxidants in oils could not reduce the total amount of PAHs effectively. Therefore, it is necessary to explore other methods to reduce PAHs in processed meat. Due to the synergistic effect of antioxidants, it would be interesting to explore replacing pork back fat with a mixture of several types of oils to reach an optimum fatty acids profile,

instead of replacing pork back fat with one single type of oil. Overall, replacing pork back fat with vegetable oils in processed meat products could offer healthier meat products with reduced HCAs without compromising eating quality.

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1127 Table 1: Formulation of control and fat reduced pork patties

Treatment	Ingredients in recipe represent as g/1kg				
	Lean pork leg	Oil	Pork back fat	Salt	Water
Control, pork back fat (C patties)	700	0	100	20	180
Fat replaced with Sunflower oil (S patties)	700	40	60	20	180
Fat replaced with Olive oil (O patties)	700	40	60	20	180
Fat replaced with Grape seed oil (G patties)	700	40	60	20	180

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1135 Table 2: Proximate composition and pH values of 4 types of patties cooked at 180 °C and 220 °C <sup>a,b</sup>.

Treatment	Cooking temperature(°C)	Moisture (%)	Cooking loss	Fat (%)	Protein (%)	pH
Control	180	69.15±0.30 <sup>d</sup>	20.51±1.59 <sup>a</sup>	10.01±0.85 <sup>a</sup>	15.19±1.56 <sup>a</sup>	5.99±0.01 <sup>a</sup>
	220	63.58±0.60 <sup>a</sup>	24.75±1.24 <sup>d</sup>	9.98±1.05 <sup>a</sup>	15.03±1.69 <sup>a</sup>	5.85±0.02 <sup>b</sup>
Olive oil	180	65.91±0.16 <sup>b</sup>	20.30±0.60 <sup>a</sup>	9.62±1.21 <sup>a</sup>	15.21±1.98 <sup>a</sup>	5.85±0.02 <sup>b</sup>
	220	63.36±0.37 <sup>a</sup>	23.21±0.84 <sup>bcd</sup>	9.49±1.25 <sup>a</sup>	15.09±1.37 <sup>a</sup>	5.86±0.01 <sup>b</sup>
Sunflower oil	180	66.34±0.63 <sup>b</sup>	22.54±0.29 <sup>bc</sup>	9.68±1.17 <sup>a</sup>	15.28±1.59 <sup>a</sup>	5.88±0.01 <sup>b</sup>
	220	63.49±0.15 <sup>a</sup>	24.23±0.76 <sup>cd</sup>	9.70±1.05 <sup>a</sup>	15.17±1.94 <sup>a</sup>	5.87±0.02 <sup>b</sup>
Grapeseed oil	180	67.53±0.26 <sup>c</sup>	21.88±0.31 <sup>ab</sup>	9.76±1.14 <sup>a</sup>	15.34±1.32 <sup>a</sup>	5.87±0.02 <sup>b</sup>
	220	65.90±0.09 <sup>b</sup>	24.72±1.09 <sup>d</sup>	9.68±1.32 <sup>a</sup>	15.17±1.46 <sup>a</sup>	5.87±0.01 <sup>b</sup>
p-value (Type of oil)		0.21	0.43	0.74	0.54	0.065
p-value (Temperature)		<0.01	<0.01	0.33	0.28	0.16
p-value (Interaction between oil* temperature)		0.18	<0.01	0.49	0.15	0.16

1136 <sup>a</sup> Results with different letters in the same column are significantly different at the level p<0.05.

1137 <sup>b</sup> Values represented as the Mean ± standard deviation (SD), n=3.

1138 Table 3: Texture parameters (hardness, cohesiveness and chewiness) and colour parameters (lightness L\*, redness a\* and  
1139 yellowness b\*) in 4 types of patties cooked at 180 °C and 220 °C <sup>a,b</sup>

Treatment	Cooking temperature (°C)	Hardness (N)	Cohesiveness	Chewiness (N.s)	Springiness	L*	a*	b*
Control	180	20.14±2.81 <sup>a</sup>	0.34±0.02 <sup>ab</sup>	4.06±0.77 <sup>ab</sup>	0.71±0.03 <sup>a</sup>	70.71±0.08 <sup>bc</sup>	2.30±0.11 <sup>c</sup>	16.98±0.80 <sup>abc</sup>
	220	26.65±3.15 <sup>b</sup>	0.37±0.01 <sup>d</sup>	5.92±0.77 <sup>c</sup>	0.77±0.04 <sup>a</sup>	65.80±1.82 <sup>a</sup>	1.46±0.22 <sup>ab</sup>	18.73±1.15 <sup>c</sup>
Olive oil	180	18.02±3.58 <sup>a</sup>	0.34±0.01 <sup>a</sup>	3.94±0.41 <sup>a</sup>	0.72±0.04 <sup>a</sup>	69.80±1.76 <sup>bc</sup>	2.82±0.08 <sup>d</sup>	16.23±0.71 <sup>ab</sup>
	220	21.75±0.37 <sup>a</sup>	0.38±0.01 <sup>d</sup>	5.35±0.41 <sup>bc</sup>	0.75±0.02 <sup>a</sup>	65.33±3.69 <sup>a</sup>	1.18±0.02 <sup>a</sup>	17.44±1.90 <sup>bc</sup>
Sunflower oil	180	17.98±2.45 <sup>a</sup>	0.35±0.01 <sup>abc</sup>	3.96±0.35 <sup>a</sup>	0.73±0.05 <sup>a</sup>	70.10±0.57 <sup>bc</sup>	2.86±0.13 <sup>d</sup>	15.41±0.73 <sup>a</sup>
	220	19.40±2.81 <sup>a</sup>	0.38±0.02 <sup>d</sup>	4.34±0.99 <sup>ab</sup>	0.74±0.03 <sup>a</sup>	65.97±2.05 <sup>a</sup>	1.43±0.22 <sup>ab</sup>	17.59±0.45 <sup>bc</sup>
Grape seed oil	180	17.86±1.12 <sup>a</sup>	0.37±0.01 <sup>cd</sup>	4.11±0.30 <sup>ab</sup>	0.77±0.01 <sup>a</sup>	71.41±1.74 <sup>c</sup>	3.24±0.37 <sup>d</sup>	17.03±0.38 <sup>abc</sup>
	220	21.55±2.37 <sup>a</sup>	0.36±0.01 <sup>bc</sup>	5.22±1.02 <sup>abc</sup>	0.77±0.04 <sup>a</sup>	68.15±0.47 <sup>ab</sup>	1.66±0.43 <sup>b</sup>	18.47±0.15 <sup>c</sup>
p-value (Type of oil)		0.10	0.87	0.50	0.341	0.062	0.54	0.65
p-value (Temperature)		0.05	0.001	0.001	0.085	0.05	0.005	0.05
p-value (Interaction between oil*temperature)		0.41	0.02	0.33	0.36	0.083	0.84	0.04

1140 <sup>a</sup> Results with different letters in the same column are significantly different at the level p<0.05.

1141 <sup>b</sup> Values represented as the Mean ± SD, n=3.

1142 Table 4: Heterocyclic amines in cooked pork patties with partial replacement of fat by vegetable oils at 180 °C and 220 °C <sup>a,b,c</sup>

Treatment	Cooking temperature	IQ (ng/g)	MeIQ (ng/g)	MeIQx (ng/g)	4,8-DiMeIQx (ng/g)	PhIP (ng/g)	Total (ng/g)	Inhibitory efficiency
Control	180	Nd	18.26±14.46 <sup>a</sup>	8.34±1.78 <sup>ab</sup>	25.66±1.51 <sup>b</sup>	11.43±6.33 <sup>a</sup>	67.56±17.29 <sup>c</sup>	N/a
	220	3.88±3.50 <sup>a</sup>	59.70±0.98 <sup>b</sup>	13.45±7.43 <sup>b</sup>	43.37±15.67 <sup>c</sup>	24.07±1.99 <sup>b</sup>	140.57±22.03 <sup>d</sup>	N/a
Olive oil	180	0.58±0.01 <sup>b</sup>	Nd	3.50±0.68 <sup>a</sup>	Nd	Nd	4.11±0.87 <sup>a</sup>	93.90%
	220	1.30±0.42 <sup>b</sup>	Nd	2.52±0.36 <sup>a</sup>	1.31±0.22 <sup>a</sup>	14.78±1.49 <sup>a</sup>	20.03±2.25 <sup>b</sup>	85.75%
Sunflower oil	180	Nd	Nd	4.32±0.50 <sup>a</sup>	1.02±0.50 <sup>a</sup>	Nd	5.98±1.10 <sup>a</sup>	91.15%
	220	0.64±0.16 <sup>b</sup>	Nd	4.31±0.55 <sup>a</sup>	5.12±0.35 <sup>a</sup>	22.70±1.95 <sup>b</sup>	23.88±2.44 <sup>b</sup>	83.01%
Grape seed oil	180	Nd	Nd	Nd	Nd	Nd	Nd	100%
	220	0.59±0.04 <sup>b</sup>	1.31±0.06 <sup>c</sup>	Nd	Nd	Nd	1.90±0.04 <sup>a</sup>	98.64%
p-value (Type of oil)		0.12	<0.01	<0.01	<0.01	<0.01	<0.01	-
p-value (Temperature)		0.037	<0.01	0.37	0.039	<0.01	<0.01	-
p-value (Interaction between oil*temperature)		0.040	<0.01	0.24	0.035	<0.01	<0.01	-

1143 <sup>a</sup> Results with different letters in the same column are significantly different at the level p<0.05.

1144 <sup>b</sup> Values represented as the Mean ± SD, n=3.

1145 <sup>c</sup> Nd: Not Detected.

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1147 Table 5: Polycyclic aromatic hydrocarbons in in cooked pork patties with partial replacement of fat by vegetable oils at 180 °C and  
1148 220 °C <sup>a,b</sup>

Treatment	Cooking temperature (°C)	BaA (ng/g)	BaP(ng/g)	Total PAHs (ng/g)	Inhibitory efficiency
Control	180	0.15±0.01 <sup>a</sup>	2.44±0.37 <sup>c</sup>	2.58±0.36 <sup>c</sup>	N/a
	220	0.21±0.03 <sup>b</sup>	3.08±0.06 <sup>d</sup>	3.28±0.07 <sup>d</sup>	N/a
Olive oil	180	0.15±0.02 <sup>a</sup>	2.24±0.40 <sup>bc</sup>	2.38±0.40 <sup>bc</sup>	7.75%
	220	0.15±0.01 <sup>a</sup>	1.44±0.27 <sup>a</sup>	1.59±0.26 <sup>a</sup>	51.52%
Sunflower oil	180	0.14±0.01 <sup>a</sup>	1.88±0.17 <sup>ab</sup>	2.02±0.16 <sup>ab</sup>	21.71%
	220	0.31±0.02 <sup>c</sup>	3.53±0.20 <sup>e</sup>	3.84±0.21 <sup>e</sup>	-17.07%
Grape seed oil	180	0.18±0.01 <sup>ab</sup>	3.29±0.15 <sup>d</sup>	3.46±0.16 <sup>d</sup>	-34.11%
	220	0.18±0.05 <sup>ab</sup>	2.51±0.07 <sup>c</sup>	2.71±0.07 <sup>c</sup>	17.38%
p-value (Type of oil)		<0.01	<0.01	0.031	-
p-value (Temperature)		0.1	0.076	0.43	-
p-value (Interaction between oil* temperature)		<0.01	<0.01	<0.01	-

1149 <sup>a</sup> Results with different letters in the same column are significantly different at the level p<0.05.

1150 <sup>b</sup> Values represented as the Mean ± SD, n=3.

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1152 Table 6: Pearson correlation coefficient (p) between the level of total HCAs/PAHs (ng/g) and TBARS, protein carbonyl and TEAC

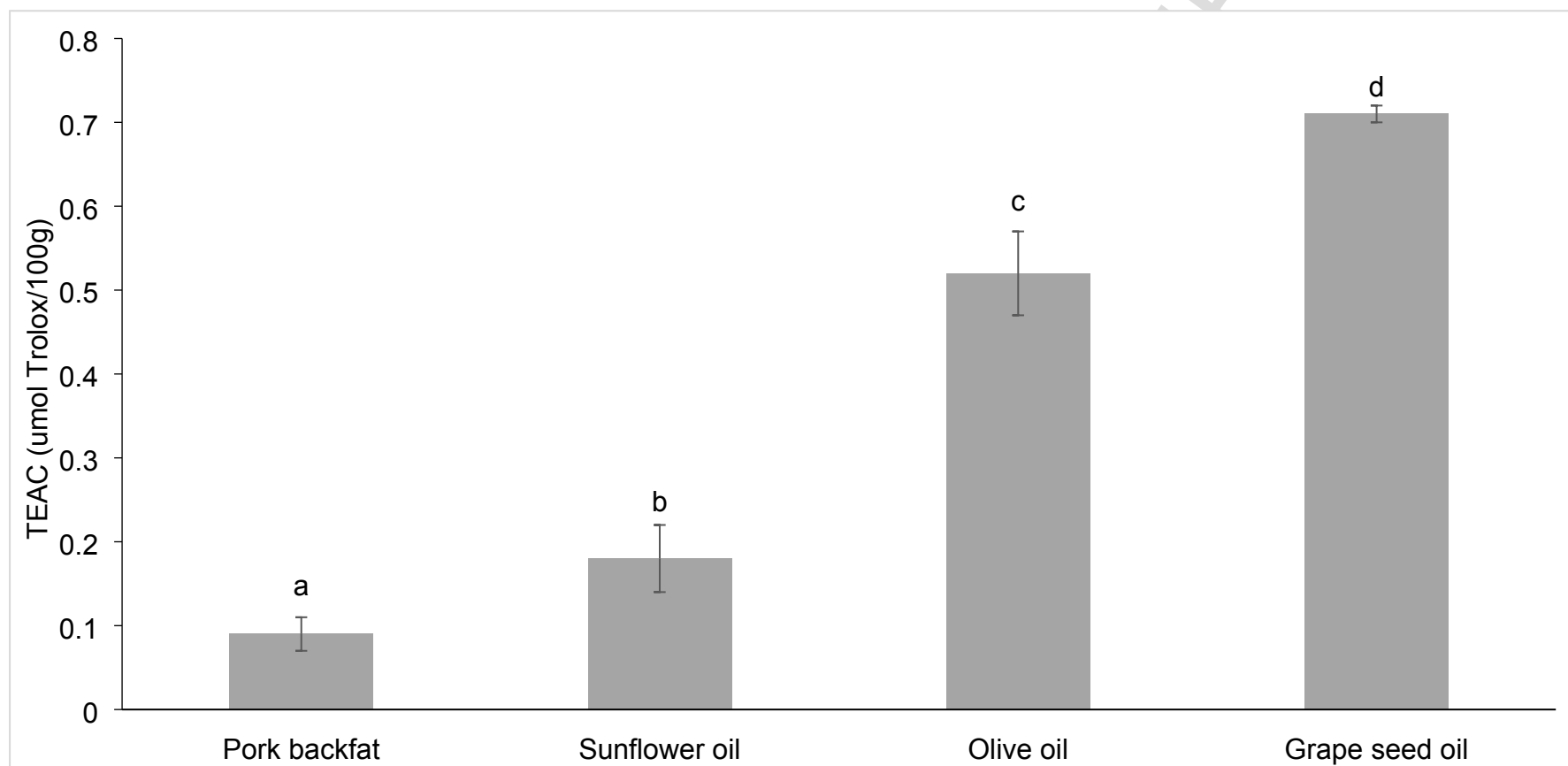
	TEAC	HCAs	PAHS
TBARS	-0.764**	0.826**	-0.154
Protein carbonyl	-0.606**	0.778**	0.019
TEAC	-	-0.618**	0.301

1153 \*\* Significant level 0.01

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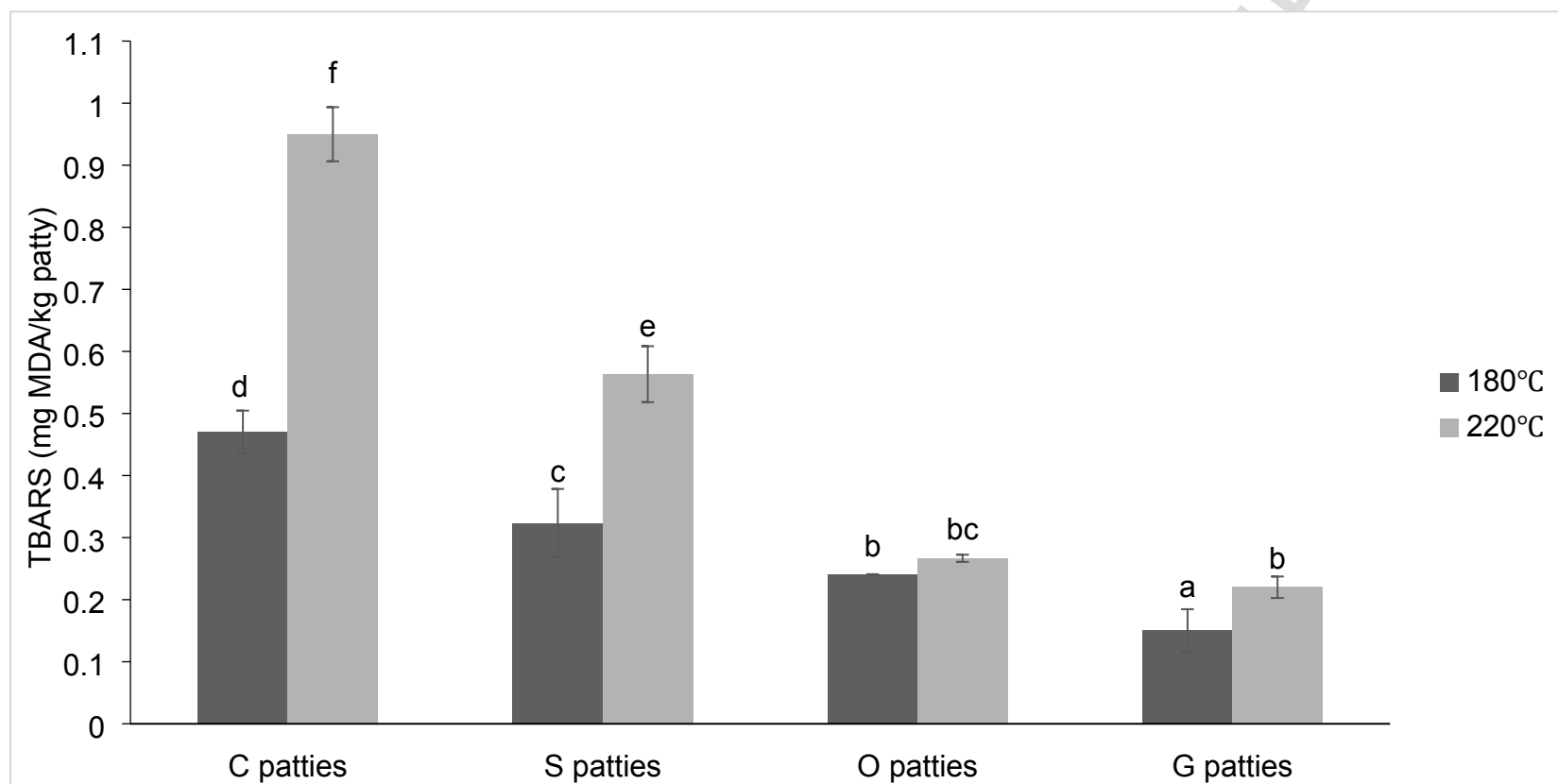


Figure 1: Trolox Equivalent Antioxidant capacity ( $\mu\text{mol Trolox}/100\text{g}$ ) of pork backfat and 3 vegetable oils.



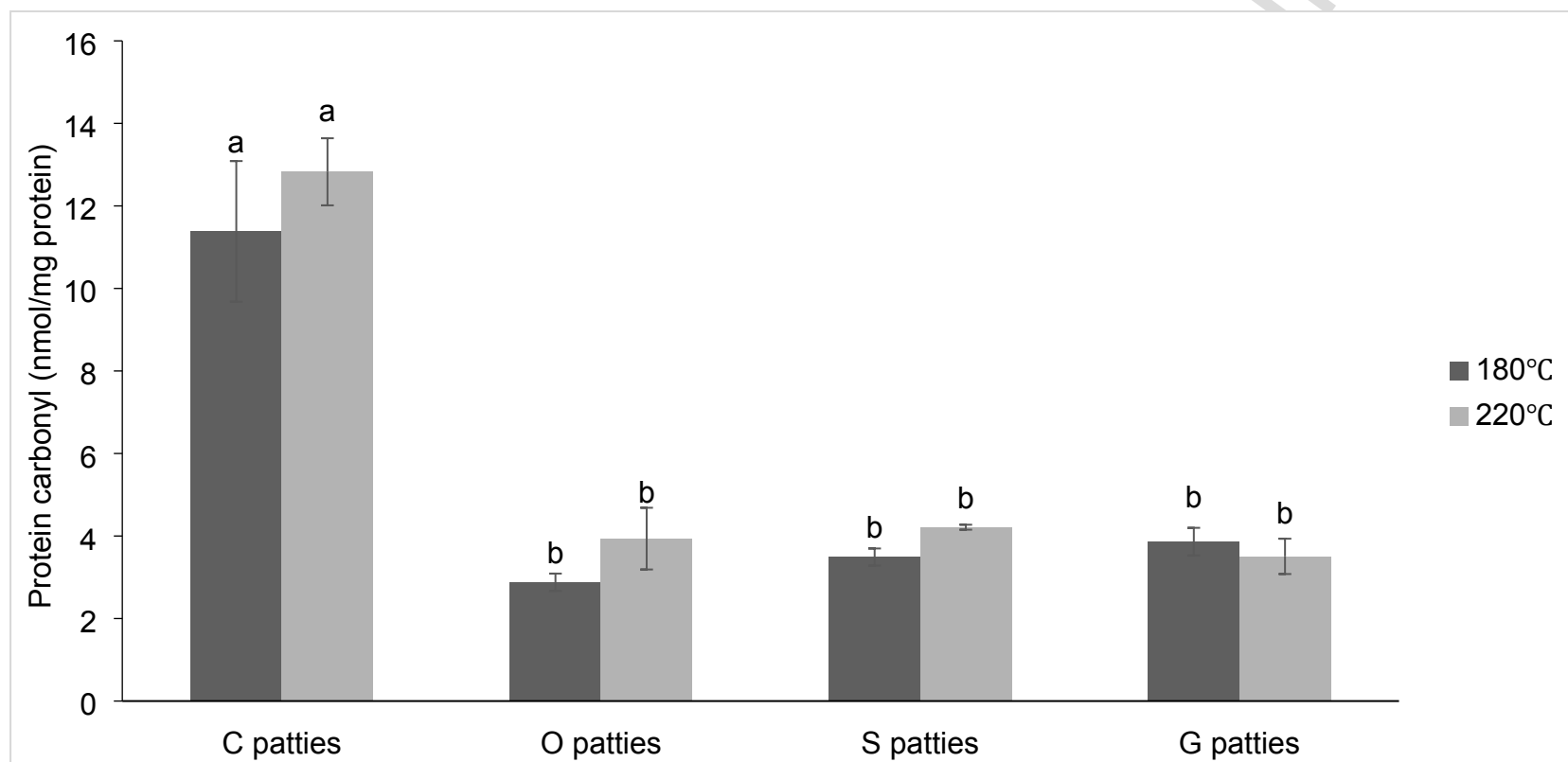
Results with different letters are significantly different at the level  $p < 0.05$ .

Figure 2: TBARS values in pork patties cooked at 180 °C and 220 °C.



Results with different letters are significantly different at the level  $p < 0.05$ .

Figure 3: Protein carbonyl values in pork patties cooked at 180 °C and 220 °C.



Results with different letters are significantly different at the level  $p < 0.05$

- Antioxidants in these oils could inhibit the formation of HCAs
- Grape seed oil achieved the highest inhibition capacity compared with sunflower oil and olive oil
- Lipid oxidation plays key role in the formation of HCAs
- Cooking temperature did not affect the total PAHs in pork patties